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3	Title	
4 5	Paper characteristics and their influence on the ability of Single Metal Deposition to detect fingermarks	
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Abstract

30	This study aims at exploring the way paper samples may impact the performance of Single-
31	Metal Deposition (SMD II), a fingermark detection technique known for its versatility of
32	application as well as its sensitivity regarding porous substrates. To get a broader view on
33	how porous substrates may impact the SMD II performances, 74 North American and
34	European papers types were collected, characterized (UV-visible and infrared spectroscopy,
35	roughness, porosity, and surface pH), and processed as substrates bearing fingermarks. This
36	part of the study represented a first valuable outcome by the number of samples considered.
37	After processing with SMD II, the samples were characterized again with the techniques
38	mentioned above, background staining and fingermark quality were assessed and associated
39	with a quality score. Overall, no positive nor negative trend was observed between the paper
40	characteristics and the SMD II performance. As a consequence, it is currently still not possible
41	to predict if a paper sample will behave well or bad with SMD II. Of all the monitored
42	parameters, the chemical composition of the surface coating (i.e., silica or calcium carbonate)
43	may be worth exploring further, as it has been observed that some coatings undergo partial
44	degradation during the SMD II process. As a result, secretion residue may be damaged by
45	the chemical solubilization of the support layer if they failed to penetrate deeper into the
46	substrate.

47	Keywords
48	Forensic Science
49	Chemical analysis
50	Porous substrate
51	Surface properties
52	Gold Nanoparticles
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54	
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56	
57	Highlights
58 59 60 61 62 63	 Physical surface topography (roughness and porosity) as well as cellulose and lignin chemical groups have no detectable influence on fingermarks detection using the SMDII technique The only factor that may be of importance seems to be the chemical composition of surface coating (silicates and carbonates).

1. Introduction

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1.1. Detection of fingermarks using single-metal deposition (SMDII)

66 Multimetal deposition (MMD) is a fingermark detection technique based on the use of colloidal 67 gold. The application protocol is built along a two-step process: (i) a detection bath containing 68 gold nanoparticles which bind to secretion residue under specific conditions, followed by (ii) a 69 contrast reinforcement bath based on the selective reduction of metal on gold nanoparticles. 70 As a result, MMD-processed fingermarks appear as dark/light-grey ridges on a relatively 71 unstained substrate [1]. Initially named "The Universal Process" [2] for its ability to detect 72 marks on a wide range of substrates (e.g., porous, non-porous, semi-porous, adhesives), the 73 technique was proposed in 1989 [3] and has consistently been improved since, to make it 74 more reliable and user-friendly. 75 Amongst the various improvements, it is possible to cite the optimization of the colloidal gold 76 synthesis, by Schnetz, to obtain more homogenous (in size and shape) and smaller (from 30 77 to 14 nm) nanoparticles [4]. This led to MMD II. Another improvement of the technique 78 consisted in replacing the silver-on-gold reinforcement step by a gold-on-gold one [5,6], that 79 proved to produce the same quality of results, with more reliable outcomes, improved control 80 and cheaper costs. At this stage of development, the technique was renamed Single-Metal 81 Deposition (SMD). Finally, the colloidal gold synthesis was further optimized, as well as the 82 application protocol to make it more end-user friendly [7,8]. The latest evolution of the 83 technique, SMD II [8], is characterized by a modified colloidal gold synthesis and a simplified 84 application protocol (e.g., no pH monitoring). As a result, the gold deposition process is more reliable and less pH dependent. 85 86 The key step of the technique (being MMD or SMD) remains the gold nanoparticles 87 deposition onto fingermark residue, which is not yet fully understood despite the various 88 optimization and improvement steps. This is a major limiting factor as it makes it difficult to 89 cope with apparent unreliability when processing items or substrates. For example, the 90 technique can give very good results on problematic substrates, such as cling films [9], but 91 suffers from several issues on conventional substrates, such as paper [10]. Among the lack of 92 reproducibility and inconsistent detection performance observed on papers, it is possible to 93 cite: unexplained background staining that can diminish the contrast, unwanted deposition of

- 94 gold nanoparticles on the substrate instead of the ridges (reversed detection), or absence of 95 detection (null result). In order to fix those issues, a better understanding of the influence 96 papers may have on the SMD performance is consequently required. 97 The main objective of this study is to monitor the effect of the composition and structure of 98 different types of paper from North American and European markets on the detection 99 efficiency of SMD II. Spectrophotometric methods as well as paper physics properties 100 (surface pH, surface profilometry, roughness and porosity) were considered to identify the 101 parameters that may influence the quality of the detected fingermarks or induce unwanted 102 background staining. Such knowledge would help designing a more robust and efficient SMD 103 formulation, so that it can be reliable independently from the types of papers. Readers 104 interested in fingermark composition and detection can refer to the most recent publications in 105 the field, such as [11]. 106 1.2. Paper composition and properties 107 1.2.1 Paper chemical composition 108 Wood represents the major raw material in the manufacture of paper, aside for specialty papers using cotton or linen, or low grade papers using annual plants. The main constituents
- papers using cotton or linen, or low grade papers using annual plants. The main constituents of wood are cellulose, hemicelluloses and lignin (Figure 1). Other components, known under the general term of extractives, are present in small and variable quantities. Two types of wood can be distinguished: softwoods (coniferous) and hardwoods, differing mostly by their content in lignin (*i.e.*, 25-35% and 18-25%, respectively). It can be noted that the lignin content in tropical hardwoods may exceed that of many softwoods. Softwoods and hardwoods share a similar amount of cellulose (40-50%), and varying structures and quantities of hemicellulose [12].
- 117 < Insert Figure 1 here >

118 **1.2.2 Cellulose**

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Cellulose is a polysaccharide consisting of a linear chain of several hundreds to many thousands of $\beta(1\rightarrow 4)$ linked D-glucose units. The cellulose macromolecules are organized in a unit called an elemental microfibril (10 nm in width and 5 nm in thickness), in which there

122 are about 100 cellulosic polymers connected by intra- and inter-molecular hydrogen bonds. 123 The main characteristic of this polymer is its insolubility in water, which is the result of the very 124 high molecular mass (3000 glucose units). 125 1.2.3 Hemicelluloses 126 Hemicelluloses differ from cellulose by the degree of polymerization (150-200), and by the 127 branching of molecular chains (Figure 1, lower right). The constitutive sugars of 128 hemicelluloses are divided into four groups: pentoses, hexoronic acids and 129 deoxyhexoses. These units are connected by $(1\rightarrow 4)$ or $(1\rightarrow 6)$ links [12]. 130 1.2.4 Lignin 131 Lignin is a thermosetting polymer with a very strong aromatic character and a molecular 132 weight that may exceed 40,000 g.mol⁻¹. The main constituting unit is the phenylpropane, 133 linked by ether-carbon or carbon-carbon bonds [12]. Lignin ensures the cohesion of the fibers 134 between each other by acting as natural glue. The complexity of lignin is such that much 135 research is still under way to define its molecular structure in a much more precise way. 136 Figure 1 shows a model of the chemical structure of softwood lignin at the top of the figure. 137 Different wood species have different lignin structure and composition. 138 1.2.5 Surface roughness 139 Paper roughness is an important parameter for its physical characterization. It is therefore 140 essential to be able to quantify the roughness of a paper so that the given value correlates 141 with the expected use of this paper, for example printing. In our case, it would be interesting 142 to see if surface roughness can be correlated with the quality of affixing of the secretion 143 residue composing the fingermarks on the different types of paper. 144 Roughness is defined as the average distance between the paper surface and a reference 145 plane to be defined. The roughness indices increase with the roughness of the paper [13]. 146 Various parameters such as Ra, Rq, Rt and Rz are defined to quantify the roughness of a 147 paper surface (Table 1). Rg is the value we will use in our analysis. 148 < Insert Table 1 here >

149 1.2.5 Porosity 150 Paper has a porous structure formed by a network of fibers. Accordingly, there is a two-phase 151 arrangement in which pores and voids between fibers form an important part of its structure 152 [14]. Paper porosity is correlated with several properties such as absorption, opacity, and ink-153 paper interaction. The porosity is influenced by the processing conditions, the addition of 154 pigments and chemical additives. For some grades of paper, a coating is applied to the top 155 surface of the paper to change its porosity [15]. 156 In forensic science, an appropriate fingermark detection sequence is usually chosen by 157 associating the item to one of the main substrate classes: porous, non-porous or semi-porous 158 (if we exclude specific substrates such as adhesives, metals, etc.). These categories are 159 based on the apparent (empiric) porosity of the substrate, which is known to influence the 160 behavior of the secretion residue [11]. Office papers are associated with porous substrates, 161 while magazine papers are usually considered as semi-porous substrates. 162 2. Material and Methods 163 2.1. Paper collection and characterization 164 2.1.1 Paper sampling 165 74 different kinds of paper (e.g., inkjet, LaserJet, copier, envelope, newsprint, Offset, drawing, 166 artistic) from 70 to 275 g.m⁻² basis weight were used in this study (see Appendix A for 167 details). These paper samples originated from Europe (e.g., Germany, Austria, Finland, 168 France, Portugal, Sweden and Switzerland), Canada, Mexico, and United States of America. 169 The samples were characterized by a range of paper composition: sugar cane (95%), 170 cardboard, colored, recycled fibers (10, 20, 30, 50% and even 80% post-consumer fibers), 171 wheat, FSC (Forest Stewardship Council) approved, Kraft, bleached Kraft, bleached generic 172 and blended mixed FSC.

2.1.2 UV-Visible-NIR spectroscopy

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174 UV-visible spectra were taken on a Varian/Agilent Cary 5000 UV-VIS-NIR spectrophotometer 175 equipped with a diffuse reflectance integrating sphere (350-850 nm for UV-Visible part and 176 4000-600 cm⁻¹ for IR). The choice of this method is necessary because paper is a solid,

177 178 179	strongly absorbent, and highly diffusive material. The main functional groups responsible for the sensitivity to light of lignin are the carbonyl and phenolic groups, quinones and various conjugated double bonds [12].
180	Ten spectra of each paper sample were recorded, averaged and analyzed using the
181	ACD/SpecManager™ version 12.00 from ACD/Labs (Advanced Chemistry Development)
182	software.
183	2.1.3 Profilometry
184	3D profiles of the paper surface were recorded with a contact free optical profilometer (Veeco
185	Wyko NT1100™ instrument) using a Mirau interferometer. Phase shift interference (PSI) and
186	vertical offset interference (VSI) can be used to respectively measure smooth and rough
187	surfaces (heights that can reach up to 1mm). Those two modes were used to optimize
188	detection and measurements of the paper samples.
189	2.1.4 Porosity measurements
190	Porosity measurements were carried out with a Parker Print Surf TM (PPS) device from
191	Hagerty Technologies™. The flow of a fluid (air in our case) that passes through the paper
192	was measured with a pressure of 1960 kPa.
193	2.1.5 Surface pH measurements
194	pH measurements were carried out with a pH Pencil from HYDRION™, measuring a gradient
195	of H ₃ O ⁺ ions on the paper surface. The first step was to moisten the surface of the paper with
196	distilled water, then to mark a line with the pen. After 15 minutes, the color of the line was
197	compared with the shades of color (color sheet) accompanying the pen. Although this method
198	is not fully accurate, it is a good way to discriminate a wide range of surface pH otherwise
199	very difficult to measure, and as pH is the most critical parameter to control for SMD
200	development, it could be planned that such an easy semi-quantitative pH tester could be
201	deployed to assist practitioners.
202	2.2 Fingermark collection

Natural and sebum-rich marks from two donors were collected for this study. For the natural fingermarks, the donors were asked not to wash their hands one hour prior deposition. No intentional enrichment was performed before collecting the fingermarks. For the sebum-rich marks, the donors rubbed their hands on their forehead before depositing the fingermarks. One natural and one sebaceous-rich marks were collected in duplicate for each donor and substrate. Fingermarks were left to age for one month in the dark. This aging period has been chosen to avoid the processing of fresh marks (e.g., one-day-old or one-week-old marks) and to focus on marks compatible with a casework timeline. Temperature and humidity were not monitored, nor controlled.

2.3. Fingermark detection, quality rating and background evaluation

The paper samples bearing fingermarks were processed using the latest SMD II protocol [8]. . Given that paper can modify the pH of the solution and have an adverse impact on the results, the paper samples were cut so that they all weight the same mass. Each paper sample was then processed in 200 ml of colloidal gold solution. Since the focus of the study is to investigate the effect of the different types of paper on the SMD II performance, each paper type was processed in a newly prepared bath of colloidal gold. After completion of the SMD II protocol, the samples were left to dry before being scanned on an Epson Perfection V330 Photo™ at 1200 dpi, without any digital enhancement. Once scanned, each mark was rated by three independent assessors using a scale ranging from 0 to 3 (Table 2 [16]).

SMD II is known to produce unwanted, uncontrolled and non-homogeneous darkening of the porous substrate. In order to understand what parameters may trigger background staining, the color of each paper was recorded before and after fingermark detection. Background measurement was done as follows: for each paper type, an unprocessed sample was placed next to a processed sample and photographed under a homogeneous lighting. Photographs were taken in grey scale and the value of the color was extracted using the eyedropper tool on Adobe Photoshop. Those values range from 0 (black) to 255 (white). For unprocessed samples, one measurement was made in the center of the paper. Processed samples required to conduct four measurements at four different locations which were then averaged, to take background staining inhomogeneity into account. The obtained value was then subtracted from the value of the unprocessed sample. A positive value means a darkening of the substrate whereas a negative value means a lightening.

234	< Insert Table 2 here >	
235	2.4. Statistical Analysis	
236 237	In order to highlight the potential correlation between the results of SMD II and the different analyses performed on paper samples, a data analysis was performed. As a first step, the	
23 <i>1</i> 238	raw results were organized using a Microsoft Office Excel spreadsheet. The analytical part	
239	was performed with the data processing software "R 3.0.2". The different methods of analysis	
240	considered were (i) the chi-square test where each variable extracted from the paper	
241	analyses was assessed against the results of SMD II (fingermark quality and background	
242	staining) and (ii) a joint analysis of variables using principal component analysis (PCA),	
243	multiple correspondence analysis (MCA) and multiple linear regression (MLR).	
244	3. Results and Discussion	
0.45		
245	3.1 Paper characterization (before SMD II)	
246	3.1.1 UV-Visible spectroscopy	
247	Figure 2 shows the processing of the UV-Visible spectrum obtained for the sample "C03".	
248	Each processed spectrum represents the variation of the log of the inverse of the reflectance	
249	as a function of wavelength.	
250	< Insert Figure 2 here >	
251	Figure 3 shows the UV-Visible spectra of some North American (C01 and C02) and European	
252	(E21 and E31) samples. The spectra show absorptions in the UV-Visible region. These	
253	absorptions are due to the presence of lignin and the colored products of the paper (dyes,	
254	coating pigments).	
255	< Insert Figure 3 here >	
256	It is possible to deconvolute the spectra to identify the electronic transitions between the	
257	various occupied molecular orbitals (OMO) and unoccupied molecular orbitals (UMO), if these	
258	are involved in our study.	

259 The 100 to 350 nm region has not been considered because of the intense background noise 260 cause by the presence of even a small amount of lignin. UV-Visible analysis of SMD II-261 processed samples shows an increase in absorption for some of the papers and a decrease 262 for others. 263 The reduction in the intensity of absorption of the spectral bands can be explained by 1) the 264 oxidation of the chromophores during the SMD II process. This has led to the displacement of 265 the other bands at longer wavelengths, towards the red part of the spectra (bathochromic 266 effect) and 2) the breaking the double bonds and formation of new compounds by 267 modification of polarity. Other papers are characterized by hypochromic displacements to 268 shorter wavelengths (towards ultraviolet). 269 3.1.2 IR spectroscopy 270 The four most important infrared regions for cellulose are the region of the bound and free OH 271 elongations between 3660 and 3000 cm⁻¹, the region of the aliphatic CH elongations between 272 3000 and 2800 cm⁻¹, the region of elongations C-O alcohols (or a cyclic system) of between 1350 and 1000 cm⁻¹, as shown in Table 3. 273 274 IR analysis showed that the many bleached papers samples designed for printing purpose 275 contain very small amounts of lignin. We also observed that about 80% of the samples (apart 276 from photographic papers and some colored papers) possess a carbonate coating, mainly 277 because these papers are intended for printing. 278 Carbonate characteristic peaks are located between 2530-2500, 1815-1770, and 1490-1370 cm⁻¹ (CO₃²⁻ elongation band), 910-850 cm⁻¹ (O-C-O deformation band), 885-870 cm⁻¹ 279 280 and 715 cm⁻¹ [17]. FTIR analysis of a calcium carbonate powder allowed the identification of 281 these bands and with the use of the ACD/SpecManager software, identification of these 282 bands in the paper samples was possible. This allowed subsequent verification of the 283 presence or loss of the carbonate layer after treatment with the SMDII. 284 < Insert Table 3 here >

285

3.1.3 Profilometry and roughness

286	The profilometry allowed obtaining the 3D topography of all paper samples (Figure 4). For	
287	sample C01, which is a glossy white paper used for inkjet printing, a uniform surface has	
288	been measured, with very low Rq (0.068±0.009µm). This makes sense given that the surface	
289	is coated with a layer of mineral pigments which makes the surface of this paper smoother	
290	and brighter.	
291	< Insert Figure 4 here >	
292	Figure 5 shows the value and variation of the Rq values of North American and European	
293	papers. The parameters Ra, Rq, Rt and Rz characterizing the roughness of each paper give	
294	the same variation, whether for North American or European papers. Rq corresponds to the	
295	quadratic mean value of the profile deviations. Rq values ranged from 0.07 \pm 0.01 μm to	
296	$6.07\pm0.01~\mu m$ for papers from North America and from $3.1\pm0.6\mu m$ to $4.5\pm1.1\mu m$ for samples	
297	from Europe. The standard deviations of the different measurements are high (Figure 5),	
298	because the samples are not uniform at the microscopic level.	
299	< Insert Figure 5 here >	
300	3.1.4 Porosity	
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313 acid-free for document preservation purposes. Only three paper types have a pH of 4 or 5 314 (C01, C20, C41). 315 3.2 Impact of SMD II on the paper properties 316 3.2.1 Carbonate-coated papers 317 The IR spectra of carbonate-coated samples before and after the application of SMD II were 318 mathematically subtracted from each other and compared with the spectrum of calcium 319 carbonate. Figure 8 illustrates the results obtained for the paper sample RetroPlus50 Canada 320 (C02). It appears that the loss observed on the spectrum resulting from the subtraction "after-321 before SMD II" (Figure 8 - top) could be correlated with the loss of calcium carbonate (Figure 322 8 - bottom). 323 < Insert Figure 8 here > This loss can be explained by the fact that SMD II requires to immerse the papers in an acidic 324 325 solution (pH close to 3). IR analysis allows identifying the peaks of the different components 326 and hence estimate a loss of carbonate. However, the quantitative analysis must be 327 completed with chemical analysis to confirm the proportions of carbonates present. When 328 assessing the % of loss by looking at the carbonate absorption band, it appears that most of 329 the paper samples loose between 20 to 50% of this layer. Some papers undergo a total loss 330 (100%). One hypothesis could be that the detrimental effect that SMD II has on the carbonate 331 layer do have a direct impact on the detection performance. This hypothesis will be 332 investigated further in this contribution. 333 3.2.2 Photographic papers 334 It was not easy to determine the exact composition of the photographic papers with the IR 335 analyzes. Figure 9 shows the three IR spectra of the sample C01, the first spectrum at the top 336 shows the result of the subtraction between the spectra recorded before and after SMD II was 337 applied. What can be seen on this difference spectrum is that there is a loss of three bands 338 which are at 1721, 1653 and 1423 cm⁻¹ and a band amplification at 1584 cm⁻¹ which is 339 identifiable in the post-SMD II spectrum. 340 < Insert Figure 9 here >

341	The subtraction between the pre-SMD II spectrum of C01 and another sample, for example
342	C04 (for which we were able to determine its cellulose and carbonate composition) shows
343	that no peak corresponds to cellulose in the C01 spectrum, as shown in Figure 10. The
344	photographic coating layer is probably too thick for the infrared radiation to reach the inner
345	layers of the paper. All the photographic papers collected in this study are also coated with
346	silicate, as shown in our FTIR spectra.
347	< Insert Figure 10 here >
348	Silicate in its various forms (gel, precipitate or colloidal) is the most widely used pigment for
349	the coating of photographic paper to provide a smooth and shiny surface [18]. The most
350	characteristic peaks of silicate are shown in Table 4 [19,20,21]. All photographic samples
351	considered in this study have a silicate coating. The positions of the spectral bands depends
352	on the type of silica used, as well as on the way the coating recipe is prepared (temperature,
353	solvent used, etc.).
354	< Insert Table 4 here >
355	Post-SMD II IR analysis showed that there was no change at the surface of the photographic
356	paper, and therefore no loss of this layer. This can be explained by the fact that the silica
357	layer is quite thick and that it remains stable at acidic pH (no dissolution of this layer). In
358	reference [18], the author describes the factors influencing the dissolution of amorphous
359	silica. Among these factors, the pH has a limited role as the dissolution is almost negligible for
360	pH below 3 and above 9. Finally, the infrared analysis did not allow the determination of the
361	type (or composition) of silica used for the coating of the photographic papers considered in
362	this study.
363	3.2.3 Colored papers
364	Some colored papers are coated with carbonate (C03, C08, C32, E30, E31, E32), while
365	others contain very little or no carbonate, such as C23 and C27 (all colors), E27, E28 and
366	E29. For the sample C27, IR analysis shows the presence of cellulose (Figure 11). The other
367	peaks belong to the aromatic compounds present (CH aromatic elongation 3083, 3060,
368	3026 cm ⁻¹ , elongation C=O 1730 and 1704 cm ⁻¹ , CC elongation of the aromatic ring at 1601,
369	1493. 1452 cm ⁻¹ , out-of-plane deformation CH aromatic at 697 cm ⁻¹).

370	For the post-SMD II IR spectrum for sample C23 (with all colors), the intensity of certain	
371	bands (1468 and 1445 and 853 cm ⁻¹) decreased, while the intensity of other bands increased	
372	(1731, 1155 cm ⁻¹). The difference between the spectra obtained before and after SMD II	
373	emphasizes this decrease but no loss of spectral bands. This is also true for sample C27.	
374	< Insert Figure 11 here >	
375	Several paper samples (C10, C33, C37, C40, E08, E11, E16, E18, E19, E20, E21, E22, E27	
376	E28, E29 E31, E30, E32) showed spectral band losses at 3340-3350 cm ⁻¹ and at 2920 cm ⁻¹ .	
377	Probably it is the CH ₂ -OH group of the cellulose which is lost during the treatment with SMD II	
378	(protonation of the alcohol in an acidic medium). The calculation of the derivative with the	
379	ACD/SpecManager software made it possible to identify the IR bands corresponding to the	
380	carbonate and the cellulose. The band at 711 cm ⁻¹ is identified in the carbonate with an	
381	accuracy of ±1 cm ⁻¹ . This band is used to calculate the amount of carbonate lost during the	
382	processing with SMD II (this band is easily identifiable) (Figure 12).	
383	< Insert Figure 12 here >	
	3.4 Correlation between the SMD II performance and the paper characteristics	
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399 observed, it appeared interesting to see if the modification of this layer may be correlated with 400 the SMD II performance. 401 < Insert Figure 13 here > 402 < Insert Figure 14 here > 403 < Insert Figure 15 here > 404 < Insert Figure 16 here > 405 When focusing on papers presenting no apparent issues in regards to fingermark detection 406 (quality scores close to 3), it can be seen that the porosity is rather low, with average airflow 407 ranging from 0 to 1000 mL/min, the roughness is characterized by Rq values ranging from ca. 408 3 to 4 microns, the surface pH of the paper ranges from 4 to 7 and the loss of carbonate 409 ranges from ca. 30% to ca. 90%. However, papers leading to bad quality fingermarks (quality 410 scores close to 0) also present similar characteristics; their porosity is low to average, with 411 average airflow ranging from 0 to 1500 mL/min, the roughness is characterized by Rq values 412 ranging from ca. 2.5 to 4 microns, the surface pH ranges from 6 to 7, and the loss of 413 carbonate ranges from ca. 10% to 100%. 414 Therefore, from the analysis of Figures 13 to 15, no clear trends can be identified regarding 415 the quality of fingermarks in regards to porosity, surface roughness or surface pH of the 416 papers analyzed. The same observation is made with the loss of calcium carbonate. 417 About the surface pH, it can be noted that papers with acidic surface pH (below 6) led to no 418 zero quality scores, which means that SMD II was able to detect fingermarks for each of 419 them, but with varying quality levels. Beyond that observation, there seems to be no trend 420 between surface pH and SMD II performance. 85% of the paper samples are indeed 421 characterized by surface pH between 6 and 7, with quality scores ranging from 0 to 3. 422 Contrarily to what could be expected, an acidic surface pH is consequently not necessarily 423 associated with better detection quality. 424 From the analysis of some 3D topographies of the different types of samples, it is remarkable 425 that the same type of surface does not give the same quality of revelation of the fingermarks,

426 as it is for the C02 samples (quality score of 0.67) and C05 (quality score of 1.84). Illustrations 427 of processed paper samples bearing fingermarks are shown in Appendix B. 428 To further refine the analysis, statistical analysis was used to try and detect correlation 429 between background noise, fingermark quality and the paper variables. However, none of the 430 techniques used (Chi-square test, PCA, MCA and MLR) led to the detection of a correlation 431 between the parameters considered. 432 3.5 General discussion 433 Despite the number of different paper types collected, the various paper properties studied 434 and the large number of fingermarks processed with SMD II, no correlation between paper 435 properties and SMD II efficiency was highlighted. However, the chemical composition of the 436 surface coating is worth discussing further. 437 Regarding the experimental design, the number of donors has been voluntarily set low 438 because the study had not for aim to assess the intrinsic performance of SMD II as 439 fingermark detection technique. It rather aimed at studying the influence of paper samples on 440 its ability to detect fingermarks. Doing so requires limiting other influencing parameters, such 441 as the variability induced by donors and the age of the fingermarks. By choosing two average 442 donors, it was possible to assess how the performance of SMD II evolves when different 443 paper samples are considered. Increasing the number of donors would have not modified the 444 overall conclusions of the study and would have imply reducing the number of paper samples 445 to keep the quantity of fingermarks manageable. 446 Surface coating is made of silica or calcium carbonate. It is used to make the surface uniform 447 and improve the printing quality [22]. This coating is however soluble in acidic aqueous 448 solutions. Therefore, immersing the samples in colloidal gold will lead to its partial dissolution. 449 If the fingermark residue does not migrate deep enough in the layer of the paper [23], it will be 450 damaged. The dissolution of the fingermark may rely on two parameters: the thickness of the 451 coating layer and the depth of penetration in the paper. According to Vallette and Choudens 452 [22], and Santos et al. [24], the thickness of the coating is about 15 µm for paper of a weight of 72 g/m² and more. It is also known that penetration depth depends on the paper type [23]. 453 454 On coated papers, observed depth could be as deep as 30 µm. It means that even if the layer

is entirely removed, a fraction of the fingermark residue may remain available for detection. Uncoated papers contain calcium carbonate as well to improve their surface characteristics and whiteness. This material may also be solubilised during SMD II processing and lead to fingermark degradation. Under these circumstances, the dynamic of diffusion of the secretion residue into the substrates is expected to play a major role. Indeed, it could be hypothesized that if the secretion residue has not migrated through the surface coating when the document is processed by SMD II, its chance of being detected would be seriously reduced. The key parameter to consider in a forthcoming study will be the aging time of the fingermarks, as we made the choice to limit our study to one-month-old fingermarks for experimental reasons. This observation is also compatible with another technique known to interact with the nonwater-soluble fraction of the secretion residue, that is, physical developer (PD). Previous studies have shown that the performance of PD increases with the age of fingermarks [25]. Moreover, this technique requires an acid pre-treatment to neutralize the alkali filler particles and to avoid an overall staining of the item. This appears compatible with the need for secretion residue to penetrate the substrate beyond the filler/coating layers to have a chance to be detected. Consequently, it may be interesting to correlate such conclusions with the results of the present study, based on SMD II.

4. Conclusions

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- This study aimed at characterizing several paper types (e.g., surface composition, surface
- 474 pH, roughness and porosity) before and after the application of SMD II. Furthermore, we
- 475 investigated the possibility to correlate the measured parameters with the performance of
- 476 SMD II, in terms of ridge quality and background staining.
- 477 At the completion of this study, we were able to show that the following parameters show no
- 478 correlation with the SMD II performance: paper roughness, porosity and surface pH. IR
- analysis showed that 81% of the papers are coated with carbonates and the thickness of this
- 480 layer varies from one sample to another. This layer appears to be solubilized during the SMD
- 481 II process. Since fingermarks are originally present at the surface of this coating, further
- 482 investigation should be carried out considering the correlation between the calcium carbonate
- 483 thickness and the SMD II detection performance. One hypothesis is that secretion residue
- 484 may migrate below the calcium carbonate layer if it is not too thick, and be further detected by
- 485 SMD II despite the dissolution of the carbonate-based coating. This hypothesis is worth being

486	further studied considering fingermarks of different ages. Moreover, it is expected that these
487	observations will be useful to physical developer as well.
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Table 1

Parameter	Description	Formula
Ra	Arithmetic mean of absolute values of deviations y	$R_a = \frac{1}{n} \sum_{i=1}^{n} y_i $
Rq	Quadratic mean value of the profile deviations	$R_q = \sqrt{\frac{1}{n} \sum_{i=1}^n y_i ^2}$
Rt	Maximum Profile Height	$R_t = R_p - R_v$
Rz	The mean height difference between the 5 highest peaks and the 5 lowest valleys	$R_z = \frac{1}{5} \sum_{i=1}^{5} R_{pi} - R_{vi}$

Table 2

Score	Qualitative observation	
0	o ridge, no fingermark visible	
1	Ridges are visible over a small area (or over the whole mark), but it is extremely difficult to retrieve level II characteristics (such as minutiae) due to extremely poor ridge details.	
2	Ridges are visible on almost the whole mark; level II characteristics can be retrieved. Nevertheless, the quality is not optimal due to a low contrast, strong background staining or faint ridges.	
3	Ridges are very well defined on the whole mark. Level II characteristics can easily be retrieved. The contrast is optimal with no (or extremely faint) background staining.	

Table 3

Frequency (cm ⁻¹)	Attribution
3332	O-H elongation with intramolecular
2897	CH ₂
1634	H ₂ O
1426	CH ₂ symmetrical deformation
1370	C-H deformation
1334	C-H shear (plane)
1316	CH ₂ agitation
1281	C-H deformation
1203	O-H deformation
1160	C-O and C-C elongation + CH ₂ rocking
1105	C-O and C-C elongation + CH ₂ rocking
1052	C-O elongation
1029	C-O elongation
1002	C-O and C-C elongation + CH ₂ rocking
897	Out-of-plane O-H deformation
659	Out-of-plane O-H deformation

Table 4

Frequency (cm ⁻¹)	Attribution
3700-3200	Si-OH
3360	H ₂ O absorbed
3000-2800	Organic C-H
1733, 1653, 1634	H ₂ O absorbed
1423	CH ₂ symmetrical deformation
1870-960	Vibrational network SiO ₂
1350-500	C-H vibration
1070	Si-O-Si symmetrical elongation
900-980	Free silanol elongation
800-820	Si-O-Si symmetrical elongation

501 Figure captions 502 Figure 1 Chemical structures of the major wood components: lignin (top), cellulose (bottom 503 left), and hemicellulose (bottom right) [12] 504 Figure 2 Example of treatment of the UV-Visible spectrum, from % Reflectance to log of 505 inverse reflectance allowing spectrum deconvolution, for the Staples Pastel (USA, 506 CO3). Red vertical line is λ_{max} , blue line is actual spectrum, green curves are the 507 resulting deconvoluted bands representing electronic transitions, and black 508 spectrum represent the result of the fitted deconvolution. 509 Figure 3 UV-Visible spectra of some of the analyzed paper samples. a) Kirkland Signature 510 (Mexico, C01), b) RetroPlus50 (Canada, C02), c) Esquisse envelope (France, 511 E21), d) Papyrus rainbow (Europe – unspecified country, E31). 512 Left half: 3D profiles of the RetroPlus50 (Canada; C02) and Staples Sustainable Figure 4 513 Earth Copy Paper (USA; C05) paper samples after they were processed with SMD 514 II. Right half: illustration of the processed samples. 515 Figure 5 Average values of the Rq parameter (µm) for all the paper samples (see Appendix 516 A for manufacturer details). 517 Average air flow (mL/min) measured for all the paper samples (see Appendix A for Figure 6 518 manufacturer details). 519 Figure 7 Chart illustrating the relation between the average airflow (mL/min) and the Rq 520 values (microns) for all the paper samples. Each dot represents a paper sample. 521 Figure 8 Top spectrum resulting from the subtraction of the IR spectra obtained before and 522 after the application of SMD II on the paper sample RetroPlus50 Canada (C02); 523 bottom IR spectrum corresponding to calcium carbonate. 524 Figure 9 Top spectrum obtained by subtracting the IR spectra obtained before (middle) and 525 after (bottom) the application of SMD II (paper sample: Kirkland Signature Mexico; 526 C01).

527	Figure 10	Difference spectra between C04 (top) and original C01 (bottom).
528 529	Figure 11	Infrared spectrum of the unprocessed Staples "Chemise à pochettes – 1336" paper sample (C27).
530 531 532	Figure 12	Derivative calculation for the RetroPlus50 paper sample (Canada; C02). Top spectra represent the paper surface with the fingermarks revealed, while bottom spectra represent the opposite surface of the same paper.
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547	Table ca	ptions
548 549	Table 1	Parameters Ra, Rq et Rz used to qualify paper surface roughness (n being the number of peaks of the profile).
550	Table 2	Table used to assess the quality of the marks (reproduced from [16]).
551 552	Table 3	Main infrared peaks characteristic of cellulose in the majority of papers studied [Erreur ! Signet non défini.].
553	Table 4	The main infrared peaks characteristic of the silicate gel [19,20,21]
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557	Competing statement
558	The authors declare that they do not have any competing interest to declare.
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563	Canada), FRQNT (Fonds de Recherche du Québec – Nature et Technologies) and CQMF
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565	support.
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Figures

Graphical Abstract (1328 x 531 pixels at 96 dpi)

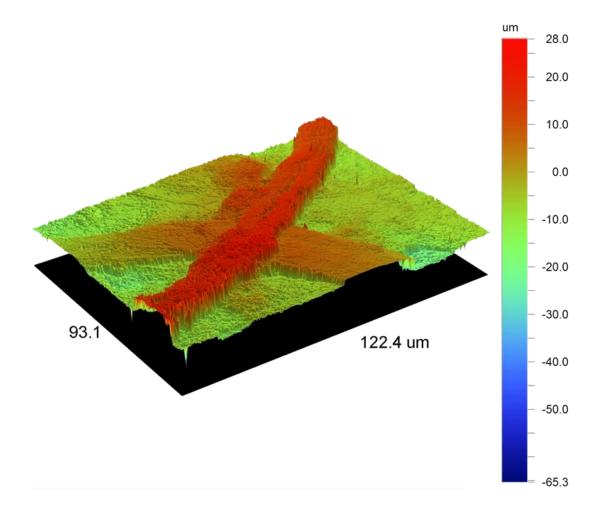


Figure Erreur! Document principal seulement. Chemical structures of the major wood components: lignin (top), cellulose (bottom left), and hemicellulose (bottom right) [12]

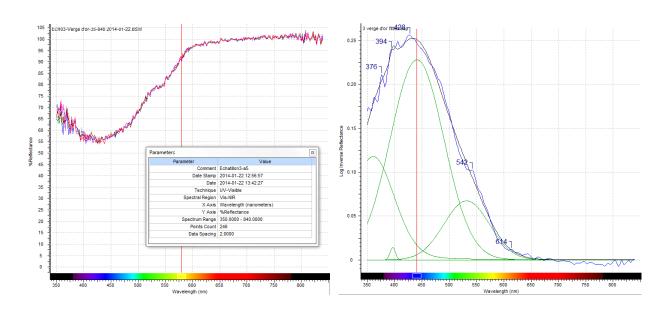


Figure Erreur! Document principal seulement. Example of treatment of the UV-Visible spectrum, from % Reflectance to log of inverse reflectance allowing spectrum deconvolution, for the Staples Pastel (USA, CO3). Red vertical line is λ_{max} , blue line is actual spectrum, green curves are the resulting deconvoluted bands representing electronic transitions, and black spectrum represent the result of the fitted deconvolution.

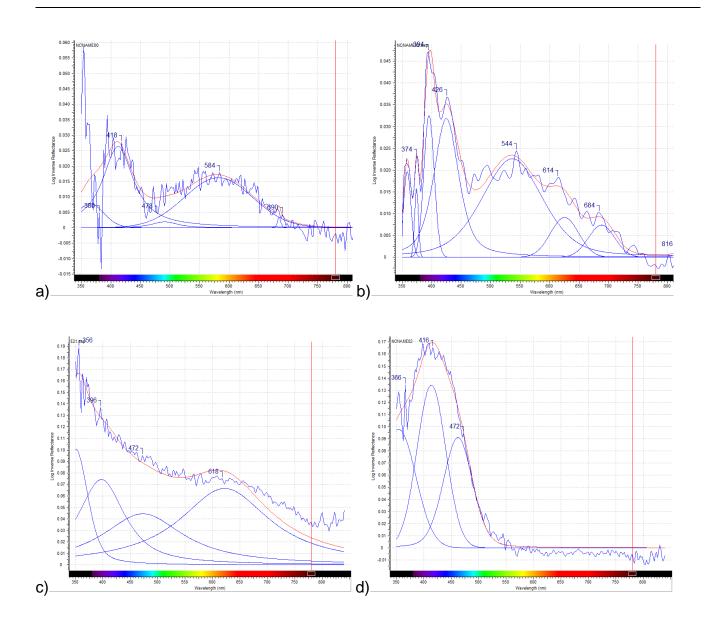


Figure Erreur! Document principal seulement. UV-Visible spectra of some of the analyzed paper samples. a) Kirkland Signature (Mexico, C01), b)
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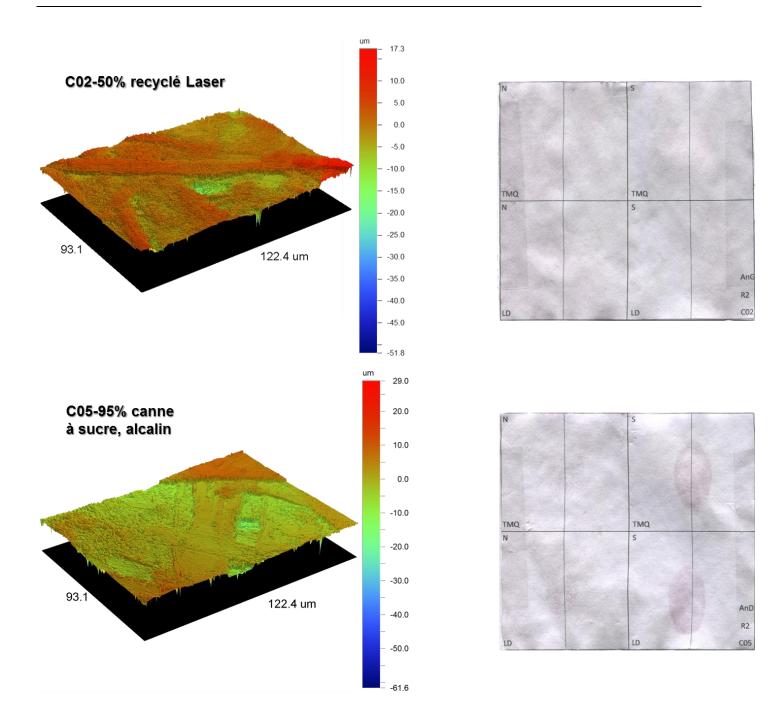


Figure 4 Left half: 3D profiles of the RetroPlus50 (Canada; C02) and Staples Sustainable Earth Copy Paper (USA; C05) paper samples after they were processed with SMD II. Right half: illustration of the processed samples.

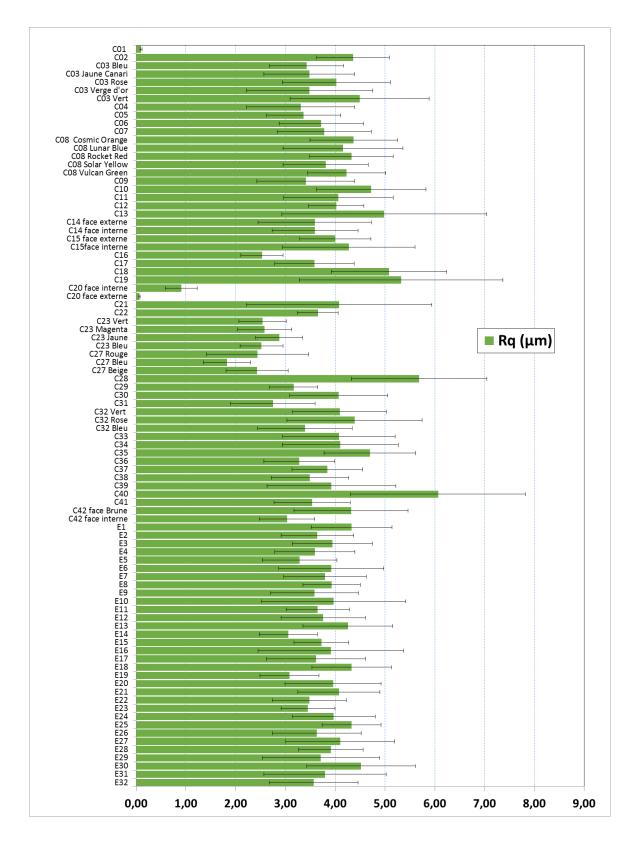


Figure 5 Average values of the Rq parameter (µm) for all the paper samples (see Appendix A for manufacturer details).

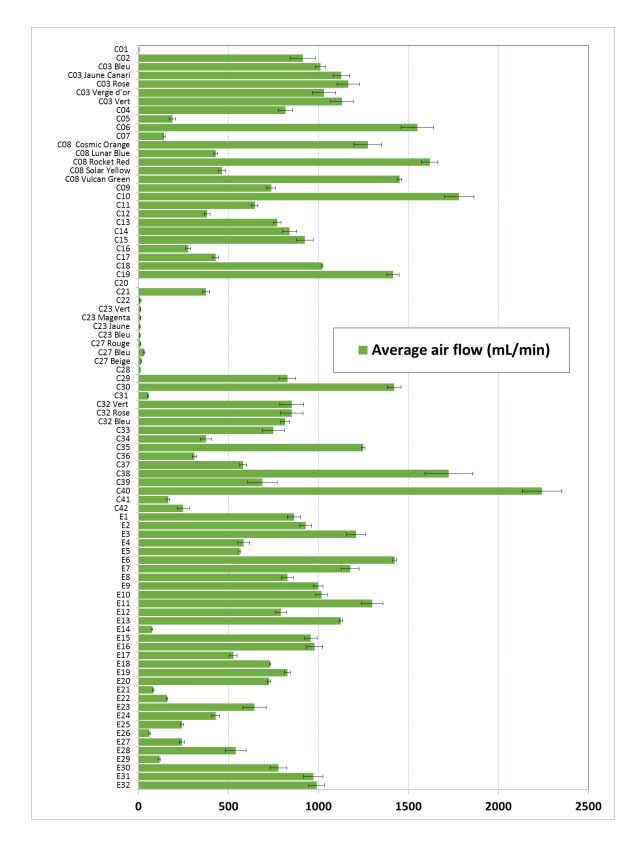


Figure 6 Average air flow (mL/min) measured for all the paper samples (see Appendix A for manufacturer details).

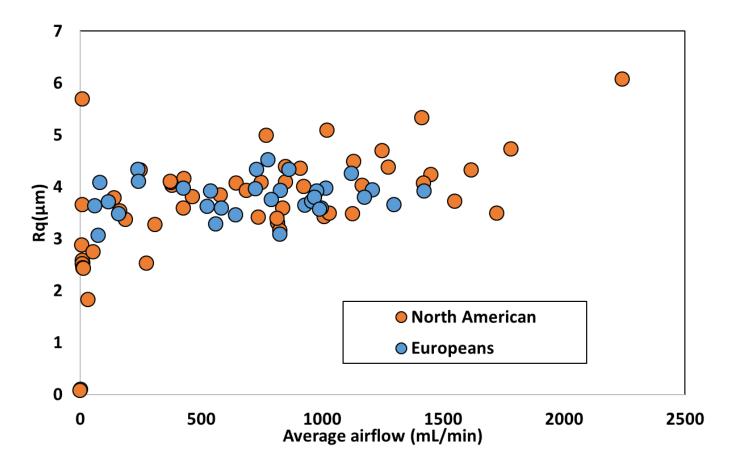


Figure 7 Chart illustrating the relation between the average airflow (mL/min) and the Rq values (microns) for all the paper samples. Each dot represents a paper sample.

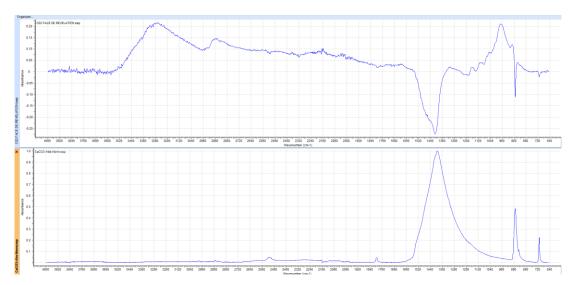


Figure 8 Top spectrum resulting from the subtraction of the IR spectra obtained before and after the application of SMD II on the paper sample RetroPlus50 Canada (CO2); bottom IR spectrum corresponding to calcium carbonate.

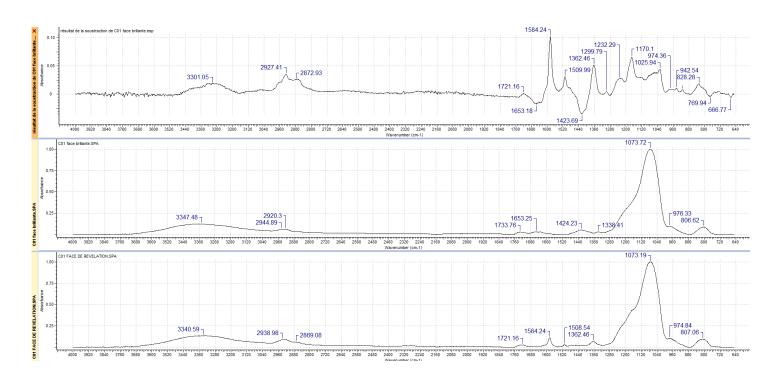


Figure 9 Top spectrum obtained by subtracting the IR spectra obtained before (middle) and after (bottom) the application of SMD II (paper sample: Kirkland Signature Mexico; C01).

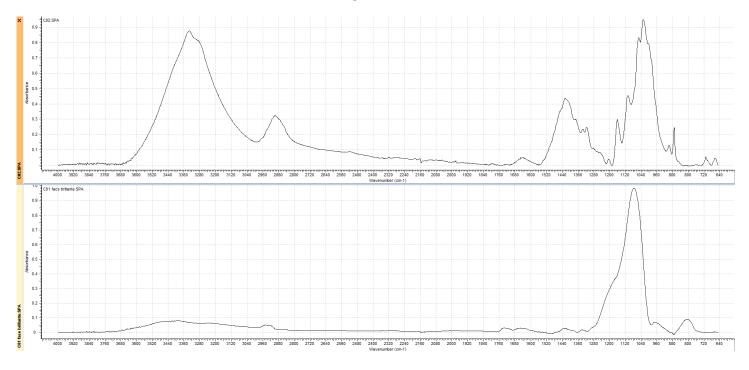


Figure 10 Difference spectra between C04 (top) and original C01 (bottom).

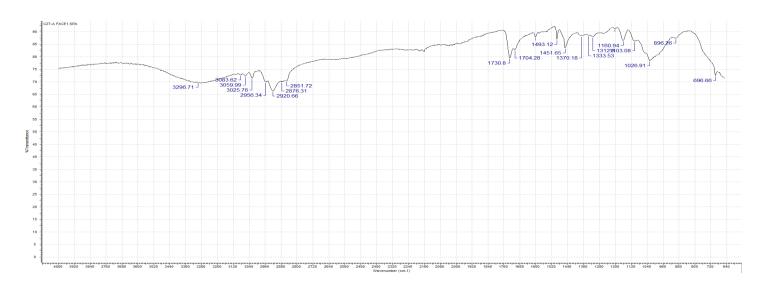


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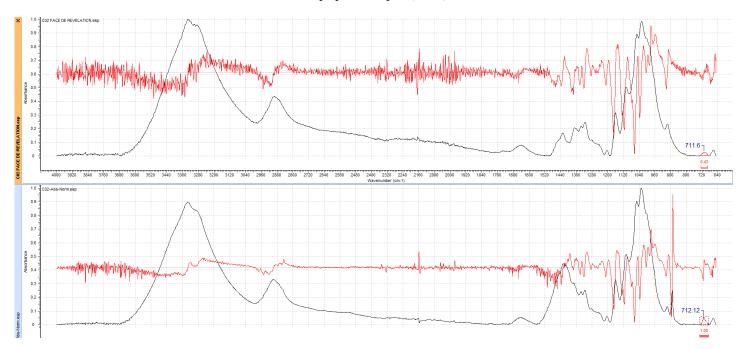


Figure 12 Derivative calculation for the RetroPlus50 paper sample (Canada; C02).

Top spectra represent the paper surface with the fingermarks revealed, while bottom spectra represent the opposite surface of the same paper.

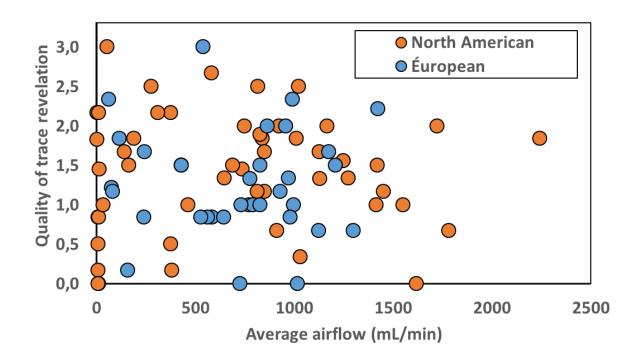


Figure 13 Chart illustrating the relation between the airflow (mL/min) and the average quality score associated with the fingermarks obtained after SMD II. Each dot represents a paper sample.

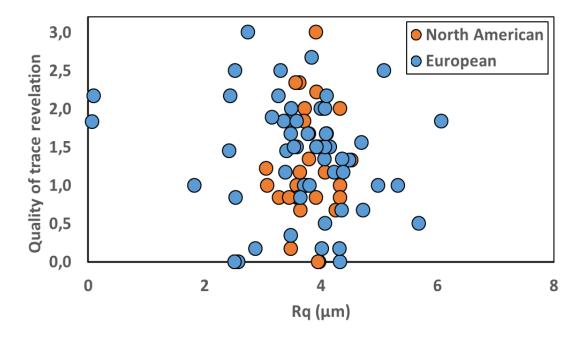


Figure **Erreur! Document principal seulement.** Chart illustrating the relation between the Rq values (microns) and the average quality score associated with the fingermarks obtained after SMD II. Each dot represents a paper sample.

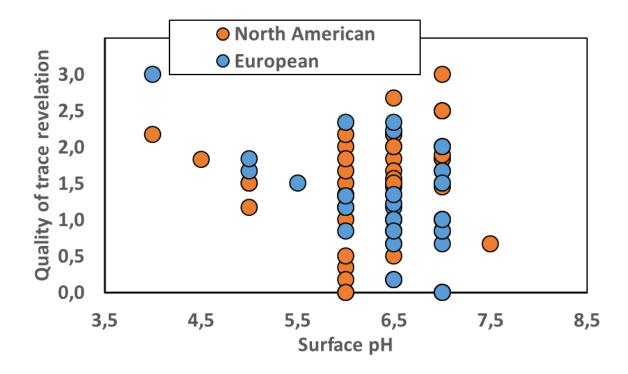


Figure Erreur! Document principal seulement. Chart illustrating the relation between the surface pH and the average quality score associated with the fingermarks obtained after SMD II. Each dot represents a paper sample.

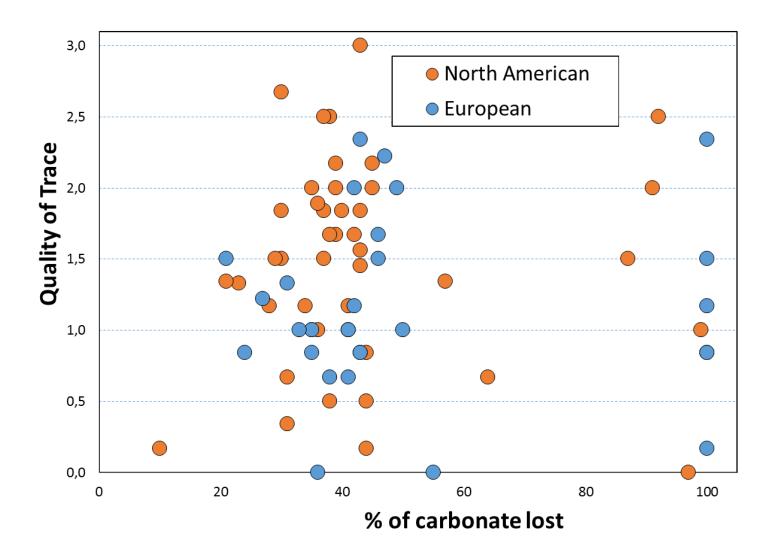


Figure 16 Chart illustrating the relation between the calcium carbonate loss (estimated %) and the average quality score associated with the fingermarks obtained after SMD II. Each dot represents a paper sample.